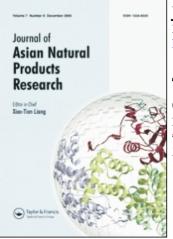
This article was downloaded by: On: 22 January 2011 Access details: Access Details: Free Access Publisher Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



### Journal of Asian Natural Products Research

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713454007

### Two new aromatic compounds from the resin of *Styrax tonkinensis* (Pier.) Craib

Feng Wang<sup>a</sup>; Hui-Ming Hua<sup>a</sup>; Xiao Bian<sup>a</sup>; Yue-Hu Pei<sup>a</sup>; Yong-Kui Jing<sup>b</sup> <sup>a</sup> Department of Natural Products Chemistry, Shenyang Pharmaceutical University, Shenyang, China <sup>b</sup> Mount Sinai School of Medicine, New York, NY, USA

**To cite this Article** Wang, Feng , Hua, Hui-Ming , Bian, Xiao , Pei, Yue-Hu and Jing, Yong-Kui(2006) 'Two new aromatic compounds from the resin of *Styrax tonkinensis* (Pier.) Craib', Journal of Asian Natural Products Research, 8: 1, 137 – 141

To link to this Article: DOI: 10.1080/10286020500480712 URL: http://dx.doi.org/10.1080/10286020500480712

# PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.



## Two new aromatic compounds from the resin of *Styrax* tonkinensis (Pier.) Craib

FENG WANG<sup>†</sup>, HUI-MING HUA<sup>†</sup>\*, XIAO BIAN<sup>†</sup>, YUE-HU PEI<sup>†</sup> and YONG-KUI JING<sup>‡</sup>

†Department of Natural Products Chemistry, Shenyang Pharmaceutical University, Shenyang 110016, China

‡Mount Sinai School of Medicine, New York, NY 10029, USA

(Received 22 June 2005; revised 30 September 2005; in final form 4 October 2005)

Two new aromatic compounds, *trans*-(tetrahydro-2-(4-hydroxy-3-methoxyphenyl)-5-oxofuran-3-yl)methyl benzoate (1), 3-(4-hydroxy-3-methoxyphenyl)-2-oxopropyl benzoate (2) and one new natural product, 4-((E)-3-ethoxyprop-1-enyl)-2-methoxyphenol (3) together with five known aromatic compounds have been isolated from the resin of *Styrax tonkinensis* (Pier.) Craib. Their structures were determined by physical and spectroscopic methods.

*Keywords: Styrax tonkinensis* (Pier.) Craib; Styracaceae; Resin; *trans*-(Tetrahydro-2-(4-hydroxyl-3-methoxyphenyl)-5-oxofuran-3-yl)methyl benzoate; 3-(4-Hydroxyl-3-methoxyphenyl)-2-oxopropyl benzoate; 4-((E)-3-Ethoxyprop-1-enyl)-2-methoxyphenol

#### 1. Introduction

*Styrax tonkinensis* (Pier.) Craib is an arbor distributed in several regions of South-East Asia and its resin has been used as a traditional medicine for expelling phlegm, waking up the patient from unconsciousness, promoting flow of Qi and blood, and alleviating pain in China. One triterpenoid, siaresinolic acid, and some aromatics series, benzoic acid, vanillin, coniferyl benzoate, cinnamyl benzoate have been isolated from the resin of the plant before [1]. In order to search the active ingredients of the resin, we do further isolation through ethanol extraction. Two new aromatic compounds **1**, **2** and one new natural product **3** were obtained together with five known aromatic compounds. The known compounds were identified as benzoic acid [2,3], vanillin [4,5], dehydrodivanillin, vanilic acid [6,7] and coniferyl aldehyde [8] (figure 1).

#### 2. Results and discussion

Compound 1 was obtained as a colorless cube. Its molecular formula was determined as  $C_{19}H_{18}O_6$  on the basis of its HREIMS spectrum (*m/z* 342.1101 [M]<sup>+</sup>). The <sup>1</sup>H-NMR spectrum gave five aromatic proton signals at  $\delta$  7.93 (2H, dd, J = 8.4, 1.8 Hz), 7.59

<sup>\*</sup>Corresponding author. Tel.: +86-24-23986465. Fax: +86-24-23892997. E-mail: huimhua@163.com

F. Wang et al.

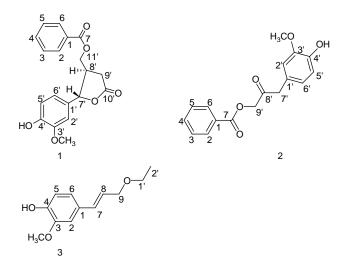


Figure 1. The structures of compounds 1, 2 and 3.

(1H, tt, J = 8.4, 1.8 Hz) and 7.44 (2H, t, J = 8.4 Hz) indicating a benzoxy group. The presence of 4-hydroxy-3-methoxyphenyl group was confirmed based on the <sup>1</sup>H-NMR signals of methoxy- and hydroxyl groups ( $\delta$  3.87 and 5.68) and three aromatic proton signals between  $\delta$  6.84 to 6.93. There were also six protons assigned to CH at  $\delta$  5.28 (1H, d, J = 7.2 Hz), CH<sub>2</sub> at  $\delta$  4.45 (2H, m), CH at  $\delta$  2.96 (1H, m) and CH<sub>2</sub> at  $\delta$  2.90 (1H, dd, J = 16.8, 8.7 Hz), 2.68 (1H, dd, J = 16.8, 8.4 Hz) in the <sup>1</sup>H-NMR spectrum. The <sup>1</sup>H-<sup>1</sup>H COSY spectrum showed correlations between CH ( $\delta$  2.96) and CH ( $\delta$  5.28), CH<sub>2</sub> ( $\delta$  4.45) and CH<sub>2</sub> ( $\delta$  2.90, 2.68), respectively. The <sup>13</sup>C-NMR spectrum showed a carbonyl group signal which was correlated with the proton signal at  $\delta$  2.68 in the HMBC spectrum. The above findings suggested the presence of a five-membered lactone.

4-Hydroxy-3-methoxyphenyl group was located at C-7' based on the HMBC spectrum which showed correlations between CH at  $\delta$  5.28 and aromatic carbons C-2' at  $\delta$  119.4 and C-6' at  $\delta$  108.2, respectively. Evidently, the benzoxy group was located at C-11'. The NOESY spectrum showed correlation between the methoxy signal at  $\delta$  3.87 and the proton signal at  $\delta$  6.84 indicating the methoxy group was located at C-3'. The *trans*-configuration of H-7' and H-8' was determined based on the presence of the correlation between H-8' and H-6' or H-2' and between H-7' and H-11'. Thus, compound **1** was identified to be *trans*-(tetrahydro-2-(4-hydroxy-3-methoxyphenyl)-5-oxofuran -3-yl)methyl benzoate.

Compound **2** was obtained as pale yellow oil. Its EIMS showed the molecular ion peak at m/z 300. Its aromatic proton signals were similar to **1** in the <sup>1</sup>H-NMR spectrum and resonated at  $\delta$  8.08 (2H, dd, J = 7.8, 1.5 Hz), 7.59 (1H, tt, J = 7.8, 1.5 Hz), 7.46 (2H, t, J = 7.8 Hz), 6.87 (1H, d, J = 8.4 Hz), 6.74 (1H, d, J = 1.8 Hz) and 6.72 (1H, dd, J = 8.4, 1.8 Hz), indicating that compound **2** also has 4-hydroxy-3-methoxyphenyl and benzoxy groups. Signals of two CH<sub>2</sub> at  $\delta$  4.93 (2H, s), 3.74 (2H, s) and one CH<sub>3</sub>O at  $\delta$  3.85 (3H, s) were shown in the <sup>1</sup>H-NMR spectrum (table 1). The <sup>13</sup>C NMR spectrum showed seventeen carbon signals including two carbonyl groups resonating at  $\delta$  201.8, 165.8, twelve aromatic carbons, two methylenes and one methoxy group (table 1). The ketone carbonyl signal at  $\delta$  201.8 in the <sup>13</sup>C-NMR spectrum correlated with the two CH<sub>2</sub> signals shown in the HMBC spectrum. Thus, the carbonyl group was located between two CH<sub>2</sub>. Furthermore, the HMBC spectrum showed the correlations between CH<sub>2</sub> at  $\delta$  4.93 and the carbonyl at  $\delta$  165.8 and between CH<sub>2</sub>

	1		2		3	
	$\delta_{H}\left(J_{Hz} ight)$	$\delta_C$	$\delta_H \left( J_{Hz}  ight)$	$\delta_C$	$\delta_{H}\left(J_{Hz} ight)$	$\delta_C$
1	_	129.2	_	129.1	_	128.2
2	7.93(1H,dd, 8.4,1.8)	129.6	8.08(1H,dd,7.8,1.5)	129.8	6.80(1H,brs)	113.5
3	7.44(1H,t,8.4)	128.5	7.46(1H,t,7.8)	128.4	_	144.7
4	7.59(1H,tt,8.4,1.8)	133.5	7.59(1H,tt,7.8,1.5)	133.4	_	144.6
5	7.44(1H,t,8.4)	128.5	7.46(1H,t,7.8)	128.4	6.74(1H,brs)	113.5
6	7.93(1H,dd,8.4,1.8)	129.6	8.08(1H,dd,7.8,1.5)	129.8	6.74(1H,brs)	119.2
7	_	166.1	_	165.8	6.40(1H,d,15.6)	131.4
8	_	_	_	_	6.01(1H,m)	122.7
9	_	_	_	_	4.02(2H,d,6.2)	70.4
1′	_	129.6	_	124.4	3.43(2H,q,6.9)	64.5
2′	6.84(1H,d,1.8)	119.4	6.74(1H,d,1.8)	122.3	1.14(3H,t,6.9)	14.1
3′	_	146.9	_	146.7	_	_
4′	_	146.2	_	145.0	_	_
5′	6.92(1H,d,8.4)	114.5	6.87(1H,d,8.4)	114.6	_	_
6′	6.86(1H,dd,8.4,1.8)	108.2	6.72(1H,dd,8.4,1.8)	111.7	_	_
7′	5.28(1H,d,7.2)	83.7	3.74(2H,s)	46.1	_	_
3'	2.96(1H,m)	43.7	_	201.8	_	_
<i>)</i> /	2.68(1H,dd,16.8,8.4) 2.90(1H,dd,16.8,8.7)	32.1	4.93(2H,s)	67.8	_	_
10′	_	175.1	_	_	_	_
11'	4.45(2H,m)	63.6	_	_	_	_
OCH <sub>3</sub>	3.87(3H,s)	56.0	3.88(3H,s)	55.9	3.72(3H,s)	54.7

Table 1	The <sup>1</sup> H-NMR and	<sup>13</sup> C-NMR data of $1, 2$ and $3$ in CDCl <sub>3</sub> <sup>a</sup> .	
Table 1.	The H-INNIK and	$C$ -INFIR that of $\mathbf{I}, \mathbf{Z}$ and $\mathbf{J}$ in $CDCI_3$ .	

<sup>a</sup> Assignment were made by <sup>1</sup>H-<sup>1</sup>H COSY, HMQC, HMBC and NOESY spectrum.

139

F. Wang et al.

at  $\delta$  3.74 and the aromatic carbons at  $\delta$  111.7, 122.3 and 124.4. These correlations suggested a fragment CH<sub>2</sub>COCH<sub>2</sub> linked to 4-hydroxy-3-methoxyphenyl at one side and formed an ester with the benzoxy group at another side. Thus, compound **2** was identified to be 3-(4-hydroxy-3-methoxyphenyl)-2-oxopropyl benzoate.

Compound **3** was obtained as a brown oil. An ethoxyl group at  $\delta$  1.14 (3H, t, J = 6.9 Hz) and 3.43 (2H, q, J = 6.9 Hz), a disubstituted olefin at  $\delta$  6.40 (1H, d, J = 15.6 Hz) and 6.01 (1H, m), a methylene group at  $\delta$  4.01 (2H, d, J = 6.2 Hz), a methoxy at 3.72 (3H, s), a hydroxy at 6.07 (1H, s) and three aromatic protons between  $\delta$  6.74 to 6.80 which indicated the presence of 4-hydroxyl-3-methoxyphenyl group were shown in the <sup>1</sup>H-NMR spectrum. The correlations between  $\delta$  3.43 (H-1') and 14.1 (C-2'), between  $\delta$  4.02 (H-9) and 64.5 (C-1'), 122.7 (C-8), between  $\delta$  6.01 (H-8) and 70.4 (C-9), 131.4 (C-7) indicating the presence of a – CH=CHCH<sub>2</sub>OCH<sub>2</sub>CH<sub>3</sub> moiety were shown in the HMBC spectrum. Moreover, the HMBC spectrum also showed a correlation between hydroxyl group at  $\delta$  6.07 and C-5 resonating at  $\delta$  113.5, which indicated that a hydroxyl group was located at C-4. Thus, compound **3** was identified to be 4-((E)-3-ethoxyprop-1-enyl)-2-methoxyphenol, which has been synthesized by A. Zanarotti [9].

#### 3. Experimental

#### 3.1 General experimental procedures

Melting points were measured on a Yanaco MP-S3 micro-melting point apparatus and are uncorrected. Optical rotations were measured with a Perkin-Elmer 241MC polarimeter. UV was recorded with Shimadzu UV-2201. NMR were recorded on a Bruker ARX 300 NMR spectrometer and a Bruker ARX 600 NMR spectrometer. The chemical shifts were quoted relative to TMS and coupling constants are in Hz. DEPT, HMBC, HMQC and COSY experiments were measured on a Bruker ARX 300 NMR spectrometer. EIMS spectra (70 ev) were conducted on a Shimadzu GCMS-QP5050A spectrometer. HREIMS was recorded on an Autospec-UltimaETOF instrument. IR was conducted on a Perkin IFS-55 spectrometer. The chromatographical silica gel (200–300 mesh) was produced by Qindao Ocean Chemical Factory and ODS (50 µm) was produced by YMC, CO., LTD.

#### 3.2 Plant material

The resin of *Styrax tonkinensis* (Pier.) Craib was bought from Liaoning Medicinal Material Corporation, Shenyang, China and identified by Prof. Qishi Sun of Shenyang Pharmaceutical University.

#### 3.3 Extraction and isolation

The resin of *Styrax tonkinensis* (Pier.) Craib (900 g) was extracted with 95% ethanol. After evaporation of ethanol, a portion (150 g) of the residue was chromatographed on a column of silica gel with a gradient elution using petroleum with increasing proportions of EtOAc and sequentially solvent gradient from EtOAc to MeOH to give ten fractions: fractions I  $\sim$  X. Fraction II [petroleum:EtOAc (100: 5)] was subsequently chromatographed over a column of silica gel to furnish compound **3** (8 mg). Repeated chromatography of fraction III

[petroleum:EtOAc (100:7)] on a column of silica gel followed by open ODS column chromatograph (70% MeOH in water) afforded compound **2** (4 mg). Fraction VI was chromatographed on a column of silica gel with gradient elution using CHCl<sub>3</sub> with increasing proportions of MeOH and was subsequently chromatographed over a column of silica gel with gradient elution using petroleum with increasing proportions of acetone to afford compound **1** (12 mg).

**3.3.1** *trans*-(Tetrahydro-2-(4-hydroxy-3methoxy-phenyl)-5-oxofuran-3-yl)mehtyl benzoate (1). Mp 136–138°C; colorless cube,  $[\alpha]_D + 7.5$  (CHCl<sub>3</sub>; *c* 0.75). UV (MeOH)  $\lambda_{max}$  215, 246 nm. IR (KBr)  $v_{max}$  cm<sup>-1</sup>: 3358 (OH), 1753 (C=O), 1720 (C=O), 1599, 1519, 1446, 1437, 1315, 1276, 716. EIMS *m/z* (rel. int): 342 [M]<sup>+</sup>(23), 220 (69), 192 (48), 151 (79). HREIMS *m/z*: 342.1101 [M]<sup>+</sup> (calcd. for C<sub>19</sub>H<sub>18</sub>O<sub>6</sub>, 342.1103). <sup>1</sup>H-NMR and <sup>13</sup>C-NMR: see table 1.

**3.3.2** 3-(4-Hydroxy-3-methoxyphenyl)-2-oxopropyl benzoate (2). Pale yellow oil. EIMS m/z (rel. int): 300 [M]<sup>+</sup>(23), 151 (71), 137 (56), 105 (100). <sup>1</sup>H-NMR and <sup>13</sup>C-NMR: see table 1.

**3.3.3 4-((E)-3-Ethoxyprop-1-enyl)-2-methoxyphenol (3)**. Pale yellow oil. <sup>1</sup>H-NMR and <sup>13</sup>C-NMR: see table 1.

#### Acknowledgements

The authors thank to Miss Wen Li and Mr. Yi Sha of the Analytical Center of Shenyang Pharmaceutical University for recording the <sup>1</sup>H-NMR, <sup>13</sup>C-NMR and 2D-NMR spectra.

#### References

- Jiangsu New Medical College. Dictionary of Chinese Materia Medica, pp. 995–996, Shanghai Scientific and Technical Publishing House, Shanghai (1993).
- [2] D. R. Lide, G.W.A. Milne. Handbook of Data on Common Organic Compound, 1, p. 505, CRC Press, USA (1995).
- [3] D. Q. Yu, J.S. Yang. *The Handbook of Analytic Chemistry*, 7, p. 70, Chemical Industry Publishing House, Beijing (1999).
- [4] D. R. Lide, G.W.A. Milne. Handbook of Data on Common Organic Compound, 1, p. 111, CRC press, USA (1995).
- [5] D. R. Lide, G.W.A. Milne. Handbook of Data on Common Organic Compound, 1, p. 110, CRC press, USA (1995).
- [6] D. R. Lide, G.W.A. Milne. Handbook of Data on Common Organic Compound, 2, p. 537, CRC press, USA (1995).
- [7] Sadtler Research Laboratories, INC, Sadtler Standard Carbon-13 NMR spectra, No. 9077.
- [8] J. Jakupovic, C. Zdero. Phytochemistry, 22, 782 (1983).
- [9] A. Zanarotti. J.Org.Chem., 50, 941 (1985).